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distribution of electrons between the oxygen and hydrogen atoms.

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hitting "snooze"
once or twice,
you pry yourself
out of bed

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chapter 2. You
make a cup of
coffee to help
you get going,
and then you
shower, get
dressed, eat
breakfast, and

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Measurements and
Calculations

SECTION 2 SHORT
ANSWER Answer
the following
questions in the

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space provided.

1. Complete the following conversions: a. 100 mL 0.1 L b. 0.25 g 25 cg c. 400 cm³ 0.4 L d. 400 cm³ 0.0004 m³
2. For each measuring device shown below, identify the quantity measured and

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tell when it
would remain
constant ...

Chapter 2

2 Measurements

and Calculations

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correct symbols

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or formulas for
the word equa

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aluminum + .

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$2(g) ? NO 2(g)$

a. Balance the
equation. $2N$

$2O(g) 3O 2(g) ?$

$4NO 2(g) 4 mol$

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NO₂:3 mol O₂

b. What is the mole ratio of NO₂ to O₂? 15.0

mol c. If 20.0 mol of NO₂

form, how many moles of O₂ must have been consumed? True

d. Twice as many moles of NO₂ form as moles of N₂O are

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Consumed. True
or False? False
e. Twice as many
grams of NO₂

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vi

CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer

the following

questions in the

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space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number.

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(d) based on
when they were
discovered. 2. d
Mendeleev
noticed that
certain
similarities in
the ...

5 The Periodic
Law

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for you to be successful. As understood, feat does not suggest that you have extraordinary points.

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2 Review
SECTION 2 SHORT
ANSWER Answer the
following

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1. Use the
concept of
potential energy
to describe how
a covalent bond
forms between
two atoms.

6 Chemical
Bonding

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